

B.Sc. Semester - 6 (CBCS) Examination

May/June-2021 (OLD COURSE)

ORGANIC CHEMISTRY AND SPECTROSCOPY(CORE)

Time: 1:30 Hours

Marks: 42

Instructions:

1. Figures to the right indicate marks.
2. There are five questions in the question paper.
3. Answer any three of the following questions.

- Q. 1 (A) Answer the following question. (04)
 (1) Explain hydantoin method with chemical reaction.
- Q.1 (B) Answer the following questions. (Any two) (10)
 (1) Explain Erlenmeyer Azlactone synthesis in detail.
 (2) Give classification of amino acid.
 (3) Explain Fischer's method for synthesis of peptide.
- Q. 2 (A) Answer the following question. (04)
 (1) Give classification of terpenoids.
- Q. 2 (B) Answer the following questions. (Any two) (10)
 (1) Write synthesis of Citral from 1,3-dibromo-3-methylbutane.
 (2) Give synthesis of (I) TNT and (II) Baygon.
 (3) Give synthesis of (I) PETN and (II) Musk Ambrette.
- Q. 3 (A) Answer the following question. (04)
 (1) Write four chemical reactions of Naphthalene.
- Q. 3 (B) Answer the following questions. (Any two) (10)
 (1) Write any five method for synthesis of biphenyl.
 (2) Explain Fischer Indole synthesis with mechanism.
 (3) Explain Skrup synthesis in detail.
- Q. 4 (A) Answer the following question. (04)
 (1) Give difference between enantiotopic and diastereotopic hydrogens in NMR.
- Q. 4 (B) Answer the following questions. (Any two) (10)
 (1) Write a note on factors affecting chemical shift in NMR.
 (2) Explain difference between chemical shift δ and coupling constant j .
 (3) Answer the following.
 a) Give two important properties of TMS.
 b) Deduce the constitution of a compound with mol formula C_2H_6O giving three signals.
 (Vapour density is 23)
 a) $3H$ Triplet 1.1 b) $2H$ Quartet 3.7 c) $1H$ singlet 2.6
- Q. 5 (A) Answer the following question. (04)
 (1) Explain Homolytic and Heterolytic fission, Base peak, molecular ion.
- Q. 5 (B) Answer the following questions. (Any two) (10)
 (1) Write a note on conformations of cyclohexane.
 (2) Explain importance of mass spectra.
 (3) Answer the following
 a) Write name of factors affecting conformations of molecule.
 b) Deduce constitution of a compound with molecular formula C_3H_6O giving following spectral properties
 UV: No absorption above 220, NMR: Strong singlet $\delta=2.1$, IR: 1720 sharp

Spectral data

NMR:	O-H	4.0,
	C-CH ₃	0.9 to 2.0 δ
	O-CH ₃	3.9-3.9 δ
	CO-CH ₃	2.4-2.7 δ
IR:	Carbonyl C=O stretching 1650-1750	
